Contribution from the Department of Chemistry and Division of Engineering, Brown University, Providence, Rhode Island 029 12, and the Department of Chemistry and Institute of Materials Science, University of Connecticut, Storrs, Connecticut 06268

# **Preparation and Properties of Iron Phosphorus Triselenide, FePSe<sub>3</sub><sup>1a</sup>**

B. TAYLOR, J. STEGER, A. WOLD,\* and E. KOSTINER<sup>1b</sup>

### *Received May* 3, *1974* AIC40287N

Iron phosphorus triselenide, FePSe,, has been prepared by chemical vapor transport using 75 Torr of chlorine gas as a transport agent and a temperature gradient of 650–>610". The structure of FePSe<sub>3</sub> (space group R3) is related to that of CdI<sub>2</sub>. Selenium atoms are in a hexagonal, close-packed array with iron atoms and phosphorus-phosphorus pairs (2: 1 ratio) occupying the trigonally distorted octahedral holes in an ordered arrangement, Magnetic susceptibility measurements show that the compound orders antiferromagnetically at 123 (2)"K; Mossbauer effect measurements indicate the principal axis of magnetization is along the crystallographic (hexagonal) c axis. The paramagnetic moment and Mossbauer data obtained above the ordering temperature indicate that the iron ions exist in the divalent high-spin state.

# Introduction

the general formula MPX<sub>3</sub>, where  $M = Mn$ , Fe, Co, Ni and  $X = S$ , Se, was first reported by Klingen, *et al.* <sup>2-5</sup> Later work on the sulfides<sup>6</sup> showed that these compounds exhibit antiferromagnetic ordering. However, there have been no reported studies of the corresponding selenide compounds. The preparation and crystal structure of compounds with

The crystal structure of  $FePSe<sub>3</sub>$  has been determined by Klingen.<sup>5</sup> The reported lattice parameters for the hexagonal unit cell (space group  $R3$ ) are  $a_0 = 6.27$  Å and  $c_0 = 19.80$  Å; theoretical density  $4.79 \text{ g/cm}^3$ ;  $Z = 6$ . The structure is related to that of  $CdI<sub>2</sub>$ , with iron atoms and phosphorus-phosphorus pairs occupying the cadmium positions and selenium atoms the iodine positions. This atomic arrangement results in FeSe<sub>6</sub> and  $P_2$ Se<sub>6</sub> octahedral groups. The P-P bond direction is collinear with the octahedral threefold axis and is parallel with the hexagonal *c* axis. The octahedral coordination about the iron atoms and theP-P pairs is shown in Figure 1 by the shaded circles.

were not definitive because this compound crystallizes in a lower symmetry monoclinic space group (but with essentially the same crystalline arrangement). The higher symmetry of the FePSe<sub>3</sub> space group allows an unambiguous interpretation of the Mossbauer data. Mossbauer effect data obtained in an earlier study of  $FePS<sub>3</sub><sup>7</sup>$ 

#### Experimental Section

cal vapor transport. The proper weights of high-purity (99.99%) starting materials were placed in silica tubes, evacuated to 2 **X**  Torr, back-filled with 75 Torr of chlorine gas, and then sealed. The chemical vapor transport reactions were carried out in a two-zone furnace. Reverse transport was carried out for 24 hr to reduce the number of nucleating centers. The furnace was then equilibrated at  $650^{\circ}$  for 3-6 hr; the final temperature gradient ( $650$ - $>$  $610^{\circ}$ ) was obtained by the gradual cooling **(l'/hr)** of the growth zone. **1.** Preparation. Single crystals of FePSe, were grown by chemi-

**2.** X-Ray Analysis. Samples prepared by the above procedures were analyzed by fast scan (1 $^{\circ}$  2 $\theta /$ min) X-ray diffractometry to en-

\* To whom correspondence should be addressed at Brown Univer sity.

**(1)** (a) This research has been supported by NSF Grant No. GH-37104 and GH-33631. Acknowledgment is also made to Brown University's Materials Sciences Laboratory, funded through the National Science Foundation-GH-33631, (b) This author's contribution was supported by the University of Connecticut Research Foundation.

(2) H. Hahn and W. Klingen, *Naturwissenschaften,* 52, 494 (1965).

- (3) (a) W. Klingen, G. Eulenberger, and H. Hahn,
- *Naturwissenschaften,* 55, 229 (1968); (b) *ibid.,* 57, 88 (1970).
- (4) W. Klingen, G. Eulenberger, and H. Hahn, paper presented at the Meeting on Transition Metal Compounds, **Oslo,** Norway, 1969. *(5)* W. Klingen, Thesis, Universitat Hohenheim, 1969.
	- (6) R. Nitsche and P. Wild, *Mater. Res. Bull.,* 5, 419 (1970).

(7) **B.** Taylor, **J.** Steger, and **A. Wold,** to be submitted for publication in *J. Solid State Chem.* 

sure that the materials were single phase and then by slow scan  $\binom{1}{4}^{\circ}$  $2\theta/\text{min}$ ) to obtain precision lattice parameters (internal standard MgO). A Norelco diffractometer with monochromatic radiation and a high-intensity copper source was used  $[\lambda(Cu K\alpha_1) 1.5405 \text{ Å}]$ . Lattice parameters were obtained from a least-squares analysis of the data.

single-crystal samples by a hydrostatic technique<sup>8</sup> using perfluoro-1methyldecalin as the liquid which was first calibrated using a crystal of high-purity silicon (density  $2.328$  g/cm<sup>3</sup>). All measurements were made at  $25 \pm 2^\circ$ . **3.** Density Measurements. Density measurements were made on

ments on ground single-crystal samples were carried out using a Faraday balance<sup>9</sup> equipped with a Cahn RG electrobalance over the temperature range **77-500°K** and at field strengths between 6.25 and 10.30 kOe. **4.** Magnetic Measurements. Magnetic susceptibility measure-

**5.** Mossbauer Effect Measurements. The iron-57 Mossbauer spectra of FePSe, were measured with a Model NS-1 Mossbauer spectrometer (Nuclear Science and Engineering Corp.) operating in the constant-acceleration mode. The 14.4-keV *7* radiation from a source of 20-mCi **"Co** diffused into Pd was detected with a gas proportional counter and collected with a 400-channel analyzer (Nuclear Chicago Corp.) operating in time-sequence scaling mode. The source and drive were calibrated against a single crystal of sodium iron(I1) nitropentacyanide dihydrate. The quadrupole splitting of this standard was taken as 1.7048 mm/sec.<sup>10</sup> Isomer shifts are reported with respect to the zero position of this standard. The data were computer fit with a maximum likelihood regression analysis program to a product of Lorentzian profiles superposed on a parabolic base line, a result of the drive geometry. All parameters-peak position, half-width and -height-were allowed to vary independently. Cryogenic measurements were made in a variable-temperature dewar (Andonian Associates, Inc.).

## Results **and** Discussion

The X-ray diffraction results from four samples of ground single crystals indicate that  $FePSe<sub>3</sub>$  has rhombohedral symmetry. The average lattice parameters (hexagonal cell) are  $a_0 = 6.262$  (3) Å and  $c_0 = 19.810$  (6) Å with a calculated density of 4.794 (3)  $g/cm<sup>3</sup>$  and an average measured density of **4.77** (1) g/cm3. These values are in good agreement with those previously reported: Table I gives the values of *hkl*  and the observed and calculated values of  $\sin^2 \theta$  for the indexed powder pattern of FePSe,.

of FePSe<sub>3</sub> over the temperature range 77-500°K. A typical plot of  $1/\chi_M$  *vs. T* (Figure 2) indicates that the material orders antiferromagnetically at 123 (2)<sup>o</sup>K. Honda-Owen  $plots<sup>11</sup>$  of all samples were made to determine the presence of any ferromagnetic impurities; none were detected in any of the samples. Magnetic susceptibility data were collected for three samples

(8) *L.* J. Cabri, *Amer. Mineral.,* 54, 539 (1969).

- (9) **B.** Morris and **A. Wold,** *Rev. Sci. Instrum.,* 39, 1937 (1968). (10) R. W. Grant, R. M. Housley, and U. Gonser, *Phys. Rev.,*
- 178, 523 (1969).
- (1 1) K. Honda, *Ann. Phys. (Leipzig),* 32 (337), 1048 (1910); M. **Owen,** *ibid.,* 37 (342), 657 (1912).



Figure 1. Schematic structure of FePSe<sub>3</sub>.



**Figure 2.** Graph of  $1/\chi_M v$ s. *T* for a ground crystal of FePSe<sub>3</sub>.





\* Only reflections used in precision lattice parameter calculations **are** shown.

The effective magnetic moment  $(P_{\text{eff}} = 5.08 \text{ (5)} \text{ BM})$  determined from a least-squares fit of the data in the paramagnetic region indicates that iron **is** present in a high-spin divalent state. Extrapolation of the inverse paramagnetic susceptibility data to zero results in a positive Weiss constant  $(\Theta = 37 (2)^{\circ} K)$ .



Figure 3. Room-temperature (296°K) Mossbauer spectra for an FePSe<sub>3</sub> single crystal as a function of the angle  $(\theta)$  between the c crystallographic axis and the incident  $\gamma$  direction: a,  $\theta = 0^{\circ}$ ; b,  $\theta =$ 45".



Figure **4.** Liquid nitrogen temperature (77" **K)** Mossbauer spectra. for an FePSe, single crystal. **as** a function *of* the angle **(0)** between the *c* crystallographic axis and the incident  $\gamma$  direction:  $a, \theta = 0^{\circ}$ ; b,  $\theta = 22.5^{\circ}$ ; c,  $\theta = 45^{\circ}$ .

Room-temperature Mossbauer spectra of FePSe, (Figure 3) were obtained for two different orientations of a single-crystal sample. The crystal was first mounted with the hexagonal c axis parallel to the incident  $\gamma$  radiation (the orientation was verified using the back-reflection Laue technique) and then reoriented so that the angle between the incident  $\gamma$  radiation and the *c* axis was 45".

For the case when the Mossbauer electric field gradient asymmetry parameter  $(\eta)$  is zero the general expressions for the relative intensities of a quadrupole-split absorption are given by eq 1 and 2,<sup>12</sup> where  $\theta$  is the angle between the direc-

$$
I_1 = 1 + \frac{3}{2}\sin^2\theta\tag{1}
$$

$$
I_2 = \frac{3}{2}(1 + \cos^2 \theta) \tag{2}
$$

tion of the  $\gamma$  radiation and the unique crystallographic axis. Since  $FePSe<sub>3</sub>$  has a rhombohedral structure with cations

**(12)** *S.* Chandra, Ph.D. **Thesis, Boston University, 1967.** 

Table **11.** Mossbauer Effect Parameters for FePSe,

$\theta$ , <sup><i>a</i></sup> deg	$T, \degree K$	$IS, b$ mm/ sec	$\Delta E_{\rm Q}$ , mm/ sec	$H$ , kOe	Ι	$I$ values		$e^2qQ/2$ ,	
						Obsd	Calcd	mm/sec	
$\mathbf 0$ 45	296(1) 296(1)	1.05(1) 1.06(1)	1.449(5) 1.441(5)		$I_1/I_2$ $I_1/I_2$	0.37(2) 0.77(2)	0.333 0.778		
0	77(1)				$I_2 + I_4$ $I_3 + I_6$	0.35(2)	0.333		
22.5	77(1)	1.19(2)		95(2)	$rac{I_1 + I_s}{I_3 + I_6}$	0.12(2)	0.105	1.53(1)	
					$I_2 + I_4$ $I_3 + I_6$	0.37(2)	0.333		
45	77(1)	1.20(2)		96(2)	$\frac{I_1 + I_5}{I_3 + I_6}$	0.48(2)	0.444	1.52(1)	
					$\underline{I_2+I_4}$ $I_3 + I_6$	0.32(2)	0.333		

 $a \theta$  is the angle between incident  $\gamma$  direction and the *c* crystallographic axis. **b** Isomer shifts are with respect to the 0 position of sodium iron(I1) nitropentacyanide dihydrate.,

on threefold axes, the observed intensities of the quadrupolesplit pair should follow eq 1 and 2. The excellent agreement between the observed and calculated intensity ratios for  $\theta =$ 0 and 45° can be seen in Table II. The isomer shift and quadrupole splitting are characteristic of high-spin Fe d<sup>6</sup> in an octahedral environment. Furthermore, the energies of the quadrupole-split lines  $(EI_1 \leq EI_2)$  indicate that the electric field gradient tensor is positive.

In agreement with the bulk antiferromagnetism observed in the magnetic susceptibility measurements, the Mossbauer spectrum of FePSe<sub>3</sub> taken at  $77^{\circ}$ K shows the presence of a magnetic hyperfine interaction (see Figure 4). The intensity equations for the hyperfine spectrum are<sup>12</sup>

$$
I_1 = I_5 = \frac{4}{3}\sin^2\theta\tag{3}
$$

$$
I_2 = I_4 = \frac{1}{3}(1 + \cos^2 \theta) \tag{4}
$$

$$
I_3 = I_6 = 1 + \cos^2 \theta \tag{5}
$$

where  $\theta$  is the angle between the principal axis of magnetization and the incident  $\gamma$  radiation.

In order to determine the correct assignment of the energy level diagram and to demonstrate that the principal axis of magnetization is parallel to the hexagonal  $c$  axis, data were collected at  $77^\circ$ K for three orientations of single-crystal FePSe<sub>3</sub>. As shown in Figure 4, the  $(I_1, I_5)$  hyperfine transitions are absent with  $\gamma$  *lc* but are apparent when  $\gamma$  is at an



Figure 5. Schematic energy level diagram for <sup>57</sup>Fe in FePSe, below the Neel point with  $\eta = 0$ ,  $H \parallel V_{zz}$ ,  $e^2 qQ > 0$ , and  $\frac{15g}{\pi}H = \frac{1}{2}e^2 qQ$ .

angle of 22.5 $^{\circ}$  to c and are even more intense when  $\gamma$  and c form an angle of 45°. The relative intensities calculated from eq 3-5 and the observed relative intensities at 0, 22.5, and 45<sup>°</sup> are in good agreement and are given in Table II. These results show clearly that the principal axis of magnetization is parallel (within experimental error) to the hexagonal c axis.

The energy level diagram determined from the positions of the various hyperfine transitions is shown in Figure 5; the internal magnetic field has a calculated value of 95 (2) kOe at 77°K.

Registry **No.** FePSe,, 52226-00-3.